## D: HANDLING OF REFRIGERANT

Because refrigerant boils at approx. $-30^{\circ} \mathrm{C}\left(-22^{\circ} \mathrm{F}\right)$ at sea level, it is cold enough to give you severe frostbite. Always wear goggles to protect your eyes and gloves to protect your hands. Also, even under the pressures normally found in CFC-12 containers, refrigerant will boil with the addition of heat. This could raise the pressure inside the container to a dangerous level.
Never expose a can of HFC-134a to direct sunlight, or to temperatures over $40^{\circ} \mathrm{C}\left(104^{\circ} \mathrm{F}\right)$. One more thing to remember about HFC-134a is that when it is exposed to an open flame or to hot metal, it forms phosgene, a deadly gas. Do not discharge HFC-134a into the atmosphere on purpose. Always read and follow the precautions on the HFC-134a bottle.


## 2. Basic Information

1) The combination of moisture and refrigerant forms acid, therefore, moisture should not be allowed to enter the refrigerant.
2) Refrigerant oil readily absorbs moisture, therefore, keep refrigerant oil containers tightly capped.
3) The process of evacuating the system is performed to remove small amounts of moisture. This is accomplished by lowering the pressure inside the system, which allows the moisture to boil off, in much the same way that a pot of water will boil away to nothing given enough time. The evacuation process does not suck the moisture out of the system.
4) A minimum level of vacuum must be reached to satisfactorily evacuate the system. This minimum level of vacuum depends on the temperature inside the system. The chart below shows the level of vacuum required to boil water at various temperatures.
Additionally, the vacuum level shown on a gauge will read approx. $3.3 \mathrm{kPa}(25 \mathrm{mmHg}, 1 \mathrm{inHg}$ ) less for each $304.8 \mathrm{~m}(1,000 \mathrm{ft})$ above sea level, due to the decrease in atmospheric pressure at altitude.

| Vacuum level required to boil water (at sea level) |  |
| :---: | :---: |
| Temperature | Vacuum |
| $1.7^{\circ} \mathrm{C}\left(35^{\circ} \mathrm{F}\right)$ | $100.9 \mathrm{kPa}(757 \mathrm{mmHg}, 29.8 \mathrm{inHg})$ |
| $7.2^{\circ} \mathrm{C}\left(45^{\circ} \mathrm{F}\right)$ | $100.6 \mathrm{kPa}(754 \mathrm{mmHg}, 29.7 \mathrm{inHg})$ |
| $12.8^{\circ} \mathrm{C}\left(55^{\circ} \mathrm{F}\right)$ | $99.9 \mathrm{kPa}(749 \mathrm{mmHg}, 29.5 \mathrm{inHg})$ |
| $18.3^{\circ} \mathrm{C}\left(65^{\circ} \mathrm{F}\right)$ | $99.2 \mathrm{kPa}(744 \mathrm{mmHg}, 29.3 \mathrm{inHg})$ |
| $23.9^{\circ} \mathrm{C}\left(75^{\circ} \mathrm{F}\right)$ | $98.5 \mathrm{kPa}(739 \mathrm{mmHg}, 29.1 \mathrm{inHg})$ |
| $29.4^{\circ} \mathrm{C}\left(85^{\circ} \mathrm{F}\right)$ | $97.2 \mathrm{kPa}(729 \mathrm{mmHg}, 28.7 \mathrm{inHg})$ |
| $35^{\circ} \mathrm{C}\left(95^{\circ} \mathrm{F}\right)$ | $95.8 \mathrm{kPa}(719 \mathrm{mmHg}, 28.3 \mathrm{inHg})$ |

